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April 23rd, 2018 - An unbalanced equation has more reactants than products or vice versa Stoichiometry the science of balancing equations uses coefficients to balance both sides of the react ? ion to follow the Conservation of Mass Ex CaH_2O gt $\text{H}_2\text{Ca}_2\text{O}_3$ This is unbalanced because there is one more calcium atom and two more oxygen atoms produced'

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