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April 12th, 2018 - Theoretical Yield Practice Problems Please go to the answer sheet to check for all answers to theoretical yield problems 1 For the balanced equation shown below if the reaction of 35.9 grams of CH₃Cl produces 6.92 grams of H₂O what is the percent yield'

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April 30th, 2018 - CH 115 SI 9 17 12 Bliss Chang 1 Session 6 Limiting Reagents Percent Yield Limiting Reagents 1 15 00 g aluminum sulfide and 10 00 g water react until the limiting reagent is used up'

'Practice Problems Limiting Reagents Answer Key

April 30th, 2018 - Practice Problems Limiting Reagents Answer Key what is the percent yield What is the limiting reagent for the formation of P 4 O 10 O 2 b' **'Answer key for percentage yield Limiting Reagents**

April 30th, 2018 - Problem One 1 For the balanced equation $C_3H_6 + 3O_2 \rightarrow 3CO + 3H_2O$ if the reaction of 21.3 grams of O_2 produces a 47.3% yield how many grams of CO would be produced'

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